
**Rubber — Nitrile latex — Determination of
residual acrylonitrile content**

*Caoutchouc — Latex de nitrile — Détermination de la teneur en
acrylonitrile résiduel*

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Foreword

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The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 3899 was prepared by Technical Committee ISO/TC 45, *Rubber and rubber products*, Subcommittee SC 2, *Testing and analysis*.

This third edition cancels and replaces the second edition (ISO 3899:1988), of which it constitutes a minor revision the main purpose of which was to update the normative references clause.

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WARNING — Persons using this International Standard should be familiar with normal laboratory practice. This standard does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user to establish appropriate safety and health practices and to ensure compliance with any national regulatory conditions.

CAUTION — Certain procedures specified in this International Standard may involve the use or generation of substances, or the generation of waste, that could constitute a local environmental hazard. Reference should be made to appropriate documentation on safe handling and disposal after use.

1 Scope

This International Standard specifies a method for the determination of the residual acrylonitrile content of nitrile rubber latices which have a residual acrylonitrile content of less than 0,2 % (by mass).

2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 123, *Rubber latex — Sampling*

ISO 1042, *Laboratory glassware — One-mark volumetric flasks*

3 Principle

A test portion of latex is distilled and the distillate collected in methanol. *n*-Dodecyl mercaptan is added to the distillate and the excess titrated with standard iodine solution.

4 Reagents

During the analysis, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

4.1 Silicone antifoaming agent, which does not affect the result of the determination.

4.2 Methanol.

4.3 Propan-2-ol.

4.4 Mercaptan solution, 1,25 % (by mass) *n*-dodecyl mercaptan solution in propan-2-ol.

4.5 Potassium hydroxide, 6 % (by mass) solution in 95 % (by volume) ethanol. The ethanol shall be free from aldehydes.

4.6 Acetic acid, glacial.

4.7 Iodine, standard volumetric solution, $c(I_2) = 0,012\ 5\ \text{mol/dm}^3$, freshly standardized with a standard reference solution of sodium thiosulfate.

5 Apparatus

Ordinary laboratory apparatus plus the following:

5.1 Distillation apparatus, consisting of a distillation flask of capacity $500\ \text{cm}^3$, a still head, a vertical water-cooled condenser and a receiver of capacity $100\ \text{cm}^3$ with a glass stopper through which pass a glass tube connected to the condenser and a shorter glass exit tube leading into a beaker of capacity $50\ \text{cm}^3$.

5.2 One-mark volumetric flask, of capacity $100\ \text{cm}^3$, conforming with the requirements of ISO 1042.

5.3 Conical flask, of capacity at least $250\ \text{cm}^3$.

6 Sampling

Sampling shall be carried out in accordance with one of the methods specified in ISO 123.

7 Procedure

7.1 Determination

Weigh $25\ \text{g} \pm 0,2\ \text{g}$ of latex into the distillation flask (see 5.1) and add $100\ \text{cm}^3$ of water and $1\ \text{cm}^3$ of the silicone antifoaming agent (4.1). Place $25\ \text{cm}^3$ of methanol (4.2) in the receiver and assemble the distillation apparatus so that the end of the tube connected to the condenser is immersed in the methanol.

Put sufficient methanol (for example $10\ \text{cm}^3$) in the beaker to cover the end of the exit tube. Immerse the receiver and beaker in ice.

NOTE The purpose of the methanol in the beaker is to collect any acrylonitrile which is not trapped by the liquid in the receiver.

Distil the mixture, adjusting the rate of boiling to control frothing, and collect $50\ \text{cm}^3$ of distillate in the receiver.

Empty the contents of the receiver and beaker into the volumetric flask (5.2). Rinse through the condenser into the receiver twice with small (for example $5\ \text{cm}^3$) portions of methanol and add the washings to the volumetric flask. Dilute to the mark with methanol.

Pipette a $50\ \text{cm}^3$ aliquot portion of the diluted distillate into the conical flask (5.3) containing $25\ \text{cm}^3$ of propan-2-ol (4.3).

Pipette $10\ \text{cm}^3$ of mercaptan solution (4.4) into the flask.

Add $1\ \text{cm}^3$ of potassium hydroxide solution (4.5) and allow to react for exactly 2 min at $23\ ^\circ\text{C} \pm 2\ ^\circ\text{C}$. Add $2\ \text{cm}^3$ of glacial acetic acid (4.6) to stop the reaction. The resulting pH should be between 4 and 6. Titrate with freshly standardized iodine solution (4.7) to a yellow colour which persists for at least 60 s. Discard the iodine solution remaining in the burette, unless it is required for immediate use.